### **RESEARCH ARTICLE**

### **CFD** evaluation of hydrophobic feedstock bench-scale fermenters for efficient high agitation volumetric mass transfer

Massachusetts, USA

Correspondence

Department of Chemical Engineering,

Dongming Xie and Seongkyu Yoon,

Email: Dongming\_Xie@uml.edu;

Seongkyu\_Yoon@uml.edu

Department of Chemical Engineering,

University of Massachusetts Lowell, 1 University Ave. Lowell, MA 01854, USA.

University of Massachusetts Lowell, Lowell,

Richard Marx 💿 | Huolong Liu | Seongkyu Yoon 💿 | Dongming Xie 💿

### Abstract

A new biomanufacturing platform combining intracellular metabolic engineering of the oleaginous yeast Yarrowia lipolytica and extracellular bioreaction engineering provides efficient bioconversion of plant oils/animal fats into high-value products. However, predicting the hydrodynamics and mass transfer parameters is difficult due to the high agitation and sparging required to create dispersed oil droplets in an aqueous medium for efficient yeast fermentation. In the current study, commercial computational fluid dynamic (CFD) solver Ansys CFX coupled with the MUSIG model first predicts two-phase system (oil/water and air/water) mixing dynamics and their particle size distributions. Then, a three-phase model (oil, air, and water) utilizing dispersed air bubbles and a polydispersed oil phase was implemented to explore fermenter mixing, gas dispersion efficiency, and volumetric mass transfer coefficient estimations  $(k_1 a)$ . The study analyzed the effect of the impeller type, agitation speed, and power input on the tank's flow field and revealed that upward-pumping pitched blade impellers (PBI) in the top two positions (compared to Rushton-type) provided advantageous oil phase homogeneity and similar estimated  $k_l a$  values with reduced power. These results show good agreement with the experimental mixing and  $k_1 a$  data.

**KEYWORDS** 

CFD, hydrophobic feedstock, multiphase flow, PBM, stirred-tank bioreactor

### 1 | INTRODUCTION

Yarrowia lipolytica, when combined with intracellular metabolic engineering and extracellular bioreaction engineering, can synthesize a wide range of valuable metabolites with U.S. FDA GRAS ("generally recognized as safe") status.<sup>[1]</sup> This makes Y. *lipolytica* an ideal choice for efficient bioconversion of oils/fats into high-value pharmaceuticals and food additives such as citric acid and wax esters.<sup>[2-5]</sup> This strictly aerobic yeast requires efficient oxygen transfer for cell growth and lipid production,<sup>[6]</sup> and oil substrates must be well-dispersed in the

Abbreviations: BC, Boundary condition; k<sub>L</sub>a, Volumetric mass transfer coefficient; MUSIG, Multiple size group; PBM, Population balance modeling; TSF, Timescale factor; vvm, Gas sparging rate (volume/min) per unit volume of unaerated liquid.

aqueous medium with high agitation for small oil droplets to attach to the surface of Y. lipolytica cells for bioconversion. Cellular and bioreaction engineering determines the overall production rate, necessitating understanding the oil particles' size, morphology, and bioreactor position (relative to the yeast). The oil's lower density, hydrophobic nature, and water insolubility necessitate strong agitation, as mixing can be the fermentation's limiting factor. This high agitation requires baffles to reduce tangential flow, whose flow impedance exacerbates the high power consumption;<sup>[7]</sup> balancing these demands makes CFD study a useful tool for designing an efficient and productive fermenter configuration.

Fermentation mixing efficiency is vital for lipid biodegradation performance.<sup>[8]</sup> Triple-impeller aerobic fermenters have demonstrated energy-efficient gas-liquid mass transfer but have liquid mixing

This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

© 2024 The Authors. Biotechnology Journal published by Wiley-VCH GmbH.

### Biotechnology

disadvantages.<sup>[9,10]</sup> Modeling the oil and air interfacial areas for mass transfer and mixing helps design these processes better.<sup>[11]</sup> Population balance modeling (PBM) has helped overcome experimental bubble size deviation<sup>[12-14]</sup> and has dramatically progressed to help predict particle sizes for high-shear and complex flow patterns of dispersed liquid-liquid systems.<sup>[12,15,16]</sup> Recently, PBMs were applied to three phases to understand particle size implications.<sup>[17]</sup> This is similar to the three phases in our fermenter, where sparging impacts the immiscible feedstock mixing.

CFD modeling has widely studied fermenters, and the high powerto-volume ratios (P/V) needed to overcome incomplete mixing or mass transfer have been well demonstrated.<sup>[9,18–20]</sup> Despite common CFD studies of fermenter gas-liquid mass transfer or liquid-liquid mixing, a gap exists in considering both simultaneously.

Despite disadvantages, Rushton-type impellers are still commonly used.<sup>[14]</sup> Pitched blade impellers were implemented in the top and middle positions to improve liquid mixing and gas holdup, targeting improved product fermentation efficiency (Liu et al., 2021). This study presents our recent modeling effort to demonstrate the inherent obstacles of sufficient hydrophobic feedstock mixing with adequate mass transfer for three impeller setups. The three-phase simulations completed with ANSYS CFX 2020 R2, which are representative of the experimental hydrodynamic parameters, are compared with key mass transfer and mixing measurements. The study found, and correlated to noteworthy experimental data, that upward-pumping pitched blade impellers in the top and middle position and a Rushton impeller on the bottom operated at 1200 RPM increased mixing and mass transfer parameters previously correlated with improved valuable metabolite synthesis.<sup>[5]</sup>

### 2 | MATERIALS and METHODS

### 2.1 | Fermenter setup

CFD simulations were based on previous Y. *lipolytica* fed-batch fermentations with vegetable oil in the medium in a 1-L working volume glass bioreactor (Sartorius Stedim UniVessel) with key geometrical details displayed in Figure 1B,C.<sup>[5]</sup> UniVessel 2-L impellers ( $3 \times 53$  mm OD evenly spaced (24 mm center-to-center)) discs were preferentially used to improve mixing. Gas holdup and bubble diameter values were determined empirically with 1 L of tap water. For oil mixing experiments, 5% (by volume) of the water was replaced with corn oil, and images were captured with a Canon EOS 6D Mark II. Experimental volumetric mass transfer coefficient values ( $k_La_e$ ) were based on the oxygen uptake rate (*OUR*) during Y. *lipolytica* fermentations controlled at a constant dissolved oxygen level ( $C_o$ ). This results in equal *OUR* and oxygen transfer rate (*OTR*) when  $C_o$  is controlled at a constant value:

$$OUR = OTR = k_L a_e \left(C_o^* - C_o\right) \tag{1}$$

where  $C_o^*$  and  $C_o$  represent the saturated (no cellular uptake) and actual dissolved oxygen concentration (with cellular oxygen uptake). The *OUR* 

was determined by an oxygen mass balance of the bioreactor while accounting for the working reactor volume:

$$OUR = \frac{\dot{m}_{in} (O_2) - \dot{m}_{out} (O_2)}{V}$$
(2)

Allowing the experimental  $k_l a$  to be estimated by:

$$k_L a_e = \frac{OUR}{(C_o^* - C_o)} \tag{3}$$

#### 2.2 | CFD geometry and mesh generation

Fluid domain booleans were created with multiple frames of reference (MFR) domains around each impeller and set with the same angular velocity. This method implicitly matches the outer, stationary domain solutions along a single boundary surface without external iterations.<sup>[21]</sup> The geometry was then symmetrically halved to reduce the size and meshed with a linear element order 1.25e-3 m tetrahedral element mesh with inflation layers added to the stationary and rotating domains near the rotating surfaces and discharge regions.<sup>[22]</sup> The mesh was then adapted to the sliding mesh (SM) approach to examine if it would better capture fermenter flow dynamics.

#### 2.3 Simulation and experimental design

CFD simulations were conducted with different impeller setups and speeds to elucidate which operational conditions may provide improved fermentation. First, simplified two-phase, oil-water (OW) simulations were completed to understand and quantify mixing by PBM of the oil phase with a free-slip wall boundary condition (BC) imposed on the liquid surface. Next, two-phase, air-water (AW) simulations were performed utilizing population balance equation-multiple size group (MUSIG) modeling to describe the air bubbling through the water with a degassing BC imposed on the liquid surface.<sup>[23]</sup> This approach allowed an average air bubble size to be calculated and implemented as a fixed-diameter air particle in the three-phase simulations. The three-phase simulations were run with the addition of a headspace and normal-speed air outlet (Figure 1A). In these three-phase simulations, water remains the continuous phase, oil is a polydispersed (MUSIG) phase, and air is a fixed-diameter dispersed phase. For all simulations, uniform yeast particle distributions are assumed due to their small size and similar density to water, allowing the suspension to be modeled with flow characteristics of the continuous phase.<sup>[24]</sup>

### 2.4 Simulation setup

Water and oil were specified for the oil/water simulations with 0.95 and 0.05 volume fractions to study mixing. The water volume fraction was 1 for the air and water simulations. For three-phase simulations, the fluid domain was extended to a 224-mm height to include the





**FIGURE 1** Pictorial of the basic simulation BCs, setup, and flow for this CFD study (A). Illustration of the 1-L working volume glass bioreactor with key dimensions noted for the PBI (B) and Rushton impeller (C) setups.

headspace, and the volume fractions were adjusted accordingly to model 1L of liquid (95% water and 5% oil by volume). The tank's walls, impellers, and baffles had no-slip conditions and utilized volume fractions to select scalable wall functions for the continuous phase for near-wall treatment. The continuous phase used k-epsilon while disperse-phased phases utilized the Dispersed Phase Zero Equation turbulence models with turbulent dispersion forces accounted for by Favre Averaged Drag Force (Dispersion Coefficient = 1). A normal speed inlet BC was specified on the ring sparger holes with an expression-calculated velocity from the specific aeration rate vvm (the ratio of air volumetric flow rate (L min<sup>-1</sup>) to the bioreactor liquid volume (L)). A similar expression was utilized to mass balance the air outlet velocity with zero specified for the oil and water velocities. A conservative timescale factor (TSF) of 1 was used for all simulations except where explicitly noted. Reduced TSFs (0.25 and 0.50) were also tested to reduce experimental deviation for high RPM simulations. For dynamic simulations, a fixed 0.001 s physical timescale was run for 120 s or until the oil reached  $\pm$  5% of the steady-state concentrations.

### 2.5 | Numerical solution

Convergence criteria of a  $1\times 10^{-5}$  root mean square (RMS) residual target was used; however, with the high turbulence in these systems, a few "hot spots" may stall residuals for valid solutions. If the RMS residual target was not met, the simulations were run on the Massachusetts

Model applied	Equations	Remarks	
Eulerian-Eulerian multiphase model	$\begin{split} & \alpha_i + \alpha_g + \alpha_o = 1\\ & \frac{\partial}{\partial t} (\rho_i \alpha_i) + \nabla (\rho_i \alpha_i \overline{u}_i) = 0\\ & \frac{\partial}{\partial t} (\rho_i \alpha_i \overline{u}_i) + \nabla (\rho_i \alpha_i \overline{u}_i \overline{u}_i) = 0\\ & \frac{\partial}{\partial t} (\rho_i \alpha_i \overline{u}_i) + \nabla (\rho_i \alpha_i \overline{u}_i \overline{u}_i) = -\alpha_i \nabla p + \nabla (\alpha_i \mu_i (\nabla \overline{u}_i + (\overline{u}_i)^T)) + \sum \widetilde{M}_{ii}^{drog} + \overline{F}_i + \alpha_i \rho_i \overline{g}\\ & F_i = -2\alpha_{F^0} \overrightarrow{\omega} \times \overrightarrow{u}_i - \alpha_i \rho_i \overrightarrow{\omega} \times (\overrightarrow{\omega} \times \overrightarrow{r}) \end{split}$	Extensive use in this type of s In isothermal bubbly flows, in multiphase-momentum eq non-drag forces have not b In the MFR method, the cons adds additional momentum	ystem.[15.23.50] terfacial momentum transfer dominates the Lations primarily affected by drag. <sup>[34]</sup> The een enabled. <sup>[23,28,51,52]</sup> cant angular velocity ( $\omega$ ) impeller rotation in the $F_i$ form.
	$\frac{\partial}{\partial t}n_{i} + \frac{\partial}{\partial t}\nabla(n_{i}\vec{u}_{i}) = P_{B} - D_{B} + P_{C} - D_{C}$ $n_{i}(t) = \frac{v_{i} + \frac{1}{2}}{\int}n(v, t)dv$ $\frac{\partial}{\partial t}(\rho_{0}\alpha_{0}f_{i}) + \Delta(\rho_{0}\alpha_{0}\vec{u}_{0}f_{i}) = S_{i}$	<ul> <li>The general form of the popu Eulerian-Eulerian model to</li> <li>Discretizing the above equation</li> <li>over the bin size dimension</li> <li>terms must equal zero.</li> </ul>	lation balance was solved along with the describe particle continuity. on into size groups and further integrating and group's mass, the summation of source
<i>k- ε</i> turbulence models	$\mu_{ m tf}=c_{\mu}~ ho((rac{k_{ m t}}{c_{ m t}})$	<ul> <li>k- ε model was applied for cor dispersed phase zero equa</li> </ul>	ttinuous phase turbulent effects. The cion was used for dispersed phases. <sup>[49,53]</sup>
	$\frac{\partial(\alpha_i\rho_ik)}{\partial t} + \nabla \left(\alpha_i(\rho_i k_i \vec{u}_i) - (\mu_i + \frac{\mu_i}{\sigma_k})\nabla k_i\right) \right) = \alpha_i \left(P_i - \rho_i \varepsilon_i\right) \\\frac{\partial(\alpha_i\rho_i\varepsilon_i)}{\partial t} + \nabla \left(\alpha_i\rho_i\varepsilon_i \vec{u}_i - (\mu_i + \frac{\mu_i}{\sigma_c})\nabla \varepsilon_i\right) = \alpha_i \frac{\varepsilon_i}{k_i} (C_{\varepsilon 1}P_i - C_{\varepsilon 2}\rho_i\varepsilon_i)$	<ul> <li>Important to note that if the l volume-based mixture valu be required to integrate th equations (Le et al., 2018); fractions, the continuous li</li> </ul>	<ul> <li>e. e were applied to all three phases,</li> <li>es for density, viscosity, and velocity would e turbulence equations into the transport however, due to the small oil and gas volume quid's turbulence is the dominating factor.</li> </ul>
Drag force	$C_D = \begin{cases} \frac{24(1+0.15R_e^{487})}{R_e}, Re \le 1000 \\ 0.44, Re > 1000 \end{cases}$	<ul> <li>C<sub>D</sub> models the complex depencent</li> <li>conditions on hydrodynam</li> </ul>	dencies of shape, inclination and flow ic bubbles/droplets (Montoya et al., 2019).
	$M_{ldp}^{aag} = -M_{dpl}^{aag} = \frac{2}{4} \frac{c_p}{d_{ab}} \alpha_{dp} \rho_l  \vec{u}_{dp} - \vec{u}_l  (\vec{u}_{dp} - \vec{u}_l)$	<ul> <li>Oil C<sub>D</sub> is calculated with the S</li> <li>(Ishii &amp; Zuber, 1979).</li> </ul>	chiller Naumann Drag correlation (Eq. 14)
	$C_{\text{D(ellipse)}} = \frac{4}{3} \frac{gd_b}{U_T^2} \frac{\rho_L - \rho_s}{\rho_l}$	<ul> <li>The Grace correlation was ap air-water systems. It accou effective diameter bubbles</li> </ul>	plied to an air bubble and was developed for nts for the geometric change of constant (Clift et al., 2005).
	$U_{T} = \frac{\mu}{\rho_{1d_{s}}} M^{-0.149}(J857)$ $J = \begin{cases} 0.94H^{0.75}, & 2 < H \le 59.3 \end{cases}$		
	$H = \frac{4}{3} E_0 M^{149} \left( \frac{\mu}{\mu_{eff}} \right)^{-0.14}$		
	$C_D = max(C_D(sphere), min(C_D(ellipse), C_D(cap))$	<ul> <li>For sparsely distributed parti particle and spherical cap li</li> </ul>	cles, CFX automatically counts the spherical mited by Equation (20).

 TABLE 1
 Mathematical models used in CFD and PBM equations.

4 of 14

(Continues)

TABLE 1 (Continued)

Model applied	Equations		Remarks
PBM	$B_{K} = .923F_{B}(1-\alpha_{0}) \left(\frac{\varepsilon_{f}}{d^{2}}\right)^{\frac{1}{2}} \int_{-\zeta}^{1} \frac{1}{(1+\varepsilon)^{2}} e^{-\left(\frac{12M_{B}\chi^{\frac{3}{2}} + 1-(\mathbf{g}_{A})^{\frac{3}{2}} - 1)\sigma}{2\rho_{f}\varepsilon^{2}/3} \frac{1}{a_{f}^{2}/\delta^{2}(1/3)}}{d\xi} d\xi$	(25)	Well-demonstrated application <sup>[12,54-58]</sup>
	$Q(V_i;V_j) = \left(F_{CT}\frac{\pi}{4}(d_i + d_j)^2 (2\varepsilon_{j}^{\frac{3}{2}} d_i^{\frac{2}{3}} + u_{qj}^2)^{\frac{1}{2}} + F_{CB}\frac{\pi}{4}(d_i + d_j)^2  U_{ij} - \sqrt{\frac{2.14\sigma}{\rho_{id_i}} + .505gd_i }\right) e^{\frac{-t_j}{q_j}}$	(26)	
	$\mathbf{t}_{ij} = \left( rac{ ho t_{ij}^2}{16\sigma}  ight)^{rac{1}{2}} \ln \left( rac{ ho_{lo}}{h_r}  ight)$	(27)	Collision efficiency is modeled with a comparison of $t_{ij}$ and actual contact time of the bound of the bo
	$\tau_{ij} = \frac{\tau_{ij}^{3,2}}{c_{c}^{2,\frac{3}{2}}}$	(28)	curre during une component; accurate results for coalescence in Euler-Euler two-fluid model with an integrated population balance model <sup>[59]</sup> but may sometimes
	$r_{ij} = 2\left(rac{1}{r_i} + rac{1}{r_j} ight)^{-1}$	(29)	over-predict coalescence, <sup>[60]</sup> which may impact mass transfer calculations.
Power	$\Gamma = \sum_{i} (\Delta \rho)_i A_i r_i$	(30)	
	$P = 2\pi\omega\Gamma$	(31)	Power numbers were derived by utilizing Ansys' built-in torque ( $\Gamma$ ) capabilities to calculate utilizing the formula in Equation (26) <sup>[61]</sup>
Uniformity	$\phi = 1 - \frac{\int_0^{\text{tr} k/0} \text{eld} \left[ \alpha_i - \alpha_i \log_i \right]}{2^{k} \int_0^{\infty} \text{eld} \left[ \alpha_i \right]}$	(32)	Uniformity indexes ( $\phi$ ) help quantitively demonstrate the reactors' dispersed phase distributions <sup>[62]</sup>

Biotechnology

5 of 14

Green High–Performance Computing Center with a 14-day run time for solutions (generally resulting in > 130,000 iterations). The local parallel calculations on the LINUX cluster were performed on 64 nodes on a 512 GB Red Hat 8 core. This run time ensured that volume fractions and  $k_L a$  values were not changing outside the quasi-steady-state solution range with additional timesteps. Multiple criteria focusing on the stabilization of critical parameters (i.e., gas holdup- or  $k_L a$ ) accompanied by reduction of residuals and energy dissipation have been previously utilized for multi-impeller systems.<sup>[25–27]</sup>

# 2.6 | Modeled equations to predict fermenter hydrodynamics and performance

Table 1 shows the CFD-modeled equations with a complete background in the supplemental information.

# 2.7 | CFD estimation of volumetric mass transfer coefficients

In combination with CFD results, Higbie's penetration theory described the gas/liquid mass transfer. This model assumes mass transfer occurs during many short and repeated gas bubble collisions, with the continuous phase generating turbulence as it continuously renews.<sup>[27]</sup> The mass transfer resistance ( $k_L$ ) was estimated by

$$k_{L} = \frac{2\sqrt{D}}{\sqrt{\pi}} \left(\frac{\rho_{L} * \varepsilon_{c}}{\mu_{l}}\right)^{\frac{1}{4}}$$
(4)

where  $\mu_l$  is the viscosity of water, and *D* is the diffusion coefficient of oxygen. The interfacial surface area of the dispersed gas phase  $(a_g)$  is calculated by

$$a_g = 6 \frac{\alpha_g}{d_{bg}} \tag{5}$$

where  $\alpha_g$  is the gas volume fraction and  $d_{bg}$  is the average air bubble size. Equations (4) and (5) were applied to an air isovolume between 0-.975 volume fractions. Successful  $k_L a$  estimations have been demonstrated for fermentation processes under relatively low agitation speeds;<sup>[14,20,26-28]</sup> however, this paper aims to provide strongly aerobic fermentation mass transfer behavior analysis with high agitation speeds, that is, up to 1200 RPM stirring speeds in a 1-L three-impeller vessel. The resulting variables were combined with a constant ( $C_k$ ) to calibrate mass transfer coefficients:

$$k_L a_{PT} = C_k k_L a_g \tag{6}$$

 $C_k$  values were determined by:

$$C_k = \frac{k_L a_e}{k_L a_g} \tag{7}$$

Biotechnology

which compares experimental yeast fermentation oxygen mass transfer ( $k_L a_e$ ) to CFD-derived estimates with the same bioreactor geometry and operating conditions. Typically, a 1-L working volume Y. *lipolytica* fermentation with three Rushton (3R) impeller setup at 1200-RPM stirring speed at 30°C, 1.0 vvm aeration, and a dissolved of oxygen of 20% air saturation ( $C_o = 0.2C_o^*$ ) has an observed OUR  $\approx$  120 mmol L<sup>-1</sup> h<sup>-1</sup>. Since the  $C_o^*$  in water at 30°C is 7.54 mg L<sup>-1</sup> or 0.24 mmol L<sup>-1</sup>, based on Equation (3), the estimated  $k_L a_e$  for these conditions is 636 h<sup>-1</sup>. Therefore,  $C_k$  helps to calibrate  $k_L a_{PT}$  to the actual bioreactor performance.

### 3 | RESULTS and DISCUSSION

### 3.1 | Two-phase (oil and water) mixing analysis

Previously,<sup>[5]</sup> two-impeller mixing analyses showed large oil particles gathering in the reactor's top center and adjacent to impeller discharges. Two three-impeller configurations (Figure 1B,C) were implemented to improve mixing and analyzed with water and 5% v/v corn oil CFD simulations for 200, 500, 1000, and 1200-RPM stirring speeds. Typical for Ruston impellers, buoyant oil particles accumulated near the shaft above the top impeller (Figure 2A)<sup>[29]</sup> at low stirrer speeds, and the higher stirrer speeds' impeller tip power caused more significant shear resulting in smaller oil droplets and better distribution.<sup>[30]</sup> The 3R simulations demonstrate that increasing stirring speed helps distribute the oil partially down the bioreactor; however, the oil tends to get "stuck" in the middle of the reactor with high-concentration pockets off the impeller discharges, limiting oil accessibility.

The top impeller's size, position, relative height to the liquid surface. and flow pattern are controlling parameters for pulling down buoyant particles.<sup>[31]</sup> By substituting two upward pumping impellers, the floating oil particles are swept to the outside and eventually drawn down (Figure 2F-H).<sup>[32]</sup> With 1000- or 1200-RPM stirring speeds, the mixing significantly improved due to stronger circulation loops developing down the reactor sides, resulting in a more homogeneous oil distribution reflected in the oil uniformity numbers. Typical for liquid-liquid dispersions in water, the average Sauter mean diameter of the oil droplets decreased with increasing system homogeneity.<sup>[12]</sup> The smaller particles and improved mixing of the oil and water with the PBI setup at 1200-RPM stirring speed have been correlated to improved Y. lipolytica fermentation for bioconversion of TAG oil into specific high-value products and closely match the improved experimental production.<sup>[5]</sup> However, two-phase oil and water simulations make simplifications, mostly the omittance of air, the impact of which will be reviewed in Section 3.3.

1-L bioreactor mixing experiments with water and 5% (v:v) corn oil were conducted to validate the oil-water mixing simulations, and the results are shown in Figure 3A. CFD mixing times were also compared, with the 3R setup having slightly longer mixing times at equivalent power inputs than the PBI setup. Experimentally, the bench-scale mixing times are almost instantaneous, and a large scale would better highlight these trends.

Figure 2 and 3A comparison demonstrates that  $\sim$ 1000 RPM is needed to utilize the reactor volume fully, and although CFD often underestimates stirred tank mixing due to turbulence model inadequacies,<sup>[23]</sup> a clear stirrer speed indication is provided. The MFR (Figure 2A-H) and SM approach (Figure 2I-P) produced similar mixing results, indicating improved mixing and homogeneity with the PBI setup. The oil phase utilized the Luo and Svendson break-up model, which model extensions<sup>[33]</sup> have highly enhanced turbulent vegetable oil-in-water emulsion CFD predictions.<sup>[34]</sup> In addition, the widely used Schiller and Naumann drag correlation only considers spherical particles and cannot correctly represent interface deformability force changes.<sup>[35]</sup> For oil and water interactions, lift was considered negligible due to the similar density and small oil droplet sizes,<sup>[12]</sup> but lift models such as the model by Frank et al.<sup>[36]</sup> should be reviewed for their larger oil droplet implications. An alternative approach to better correlate CFD with experimental results would be to tune  $F_B^{[25]}$  encouraging easier oil break-up and diffusion. However, for longer fermentation runs, the oil particles adhere to the shaft, baffles, and impellers (like the CFD models), indicating that the process is in multiple ways well-characterized by the steady-state, oil-water CFD modeling.

## 3.2 | Two-phase (air and water) particle size characterization

CFD modeling next examined the vessel's gas holdup and air particle size distributions with the MFR and SM approach utilizing air and water with a degassing BC. The average air bubble size is a crucial derived parameter to help provide a similar gas interfacial area to an air PBM while using a computationally less-intensive dispersed phase for the three-phase models. The CFD models indicate that the average air bubble size decreases as power input increases due to increased shear and turbulence. Average air bubble sizes displayed in Figure 3B were completed without oil, which clouds the reactor, hindering photographic bubble size verification. Experimental high-speed camera images were compared to the average bubble sizes to ensure reasonable agreement. High-speed camera images for 200-, 500- and 1200-RPM stirring speeds are displayed in Figure 3B (1000-RPM stirring speeds omitted due to 1200-RPM similarity) and compared to the average particle size (to-scale) for reference (Figure 3B).

Figure 4A compares CFD versus Xie et al. correlation-derived gas holdup values, which generally show good agreement except for the 3R 1000 and 1200 RPM results.<sup>[37]</sup> The holdup discrepancies cannot be solely attributed to some known lab-scale MFR inaccuracies for high impeller-to-tank diameter ratios,<sup>[38]</sup> as the SM approach also displays significant experimental holdup deviation, which might be expected without turbulent dispersion force and drag coefficient modifications.<sup>[39]</sup> There appear to be two contributing factors: Power input impacts the simulation's accuracy as simulations with P/Vs less than 6,000 W m<sup>-3</sup> provide reasonable holdup results. Secondly, the degassing BC may ignore close-to-surface, top-impeller, high-velocity, impeller-shaft vortexes that trap air in the reactor.<sup>[38]</sup> The degassing



**FIGURE 2** Oil volume fraction contours for MFR 3R (A–D) and PBI (E–H) and SM approach 3R (I–L) and PBI (M–P) impeller setups in the 1-L working volume bioreactor.



**FIGURE 3** Experimental oil mixing comparison between oil flow pattern development with PBI (left) and 3R (right) impeller setups (A). Experimental validation bioreactor runs with the CFD average air particle sizes imposed into the pictures for 3R (top row) and PBI (bottom row) impeller setups (B). The 1000-RPM stirring speed runs were excluded due to their similarity (pictorially and particle size similarity) to the 1200-RPM stirring speed runs. The liquid volume was 1 L for all experiments/simulations and images were captured with a Canon EOS 6D Mark II .



**FIGURE 4** Comparison of air/water CFD simulations to the Xie et al. correlation gas holdup<sup>[37]</sup> as a function of power input (P/V) (A). Experimental and air/water CFD simulation gas holdup comparison as a function of power input (P/V) (B). Three-phase (headspace included) experimental and CFD gas holdup as a function of power input (P/V)<sup>[37]</sup> (C). The CFD calculated  $C_k k_L a_{PT}$  as a function of P/V for three-phase (air, oil, and water) simulations. The liquid volume was 1 L for all simulations.

BC acts as a water-free-slip wall and air outlet, disallowing headspace air reentry. Instead, a pressure distribution based on flow surface height variations is used, with a fixed reference pressure point automatically set for the domain. Ignoring these vortexes and the inability for surface air drawdown appears to contribute to the degassing BC showing reduced CFD holdup for these high-agitation Rushton simulations. Reducing the TSF for MFR models only modestly (~10%) reduced the CFD-experimental discrepancy. This lower Rushton holdup could indicate potential impeller flooding evinced by poorly dispersed gas that rises directly along the shaft, or high impeller speeds may cause vortices behind the impeller blades, dispersing the air<sup>[43]</sup> but require a more detailed study of the phenomenon.

Experimentally, the 3R and PBI setups produce similar low stirring speed gas holdup, with the PBI setup showing better energy efficiency for similar holdup values above 5,000 W m<sup>-3</sup> (Figure 4B). This more efficient holdup may indicate that the PBI setup reaches effective gas dispersion limited by the bubble break-up/coalescence equilibrium with less power than the 3R.<sup>[10]</sup> These findings are consistent with multiple studies noting higher gas holdup utilizing upward-pumping in the top two positions and a radially dispersing impeller on the bottom but are limited to lower P/V ratios (P/V < 4000 W m<sup>-3</sup>)<sup>[37,40,41]</sup> with similar two-impeller trends.<sup>[42]</sup> With the noted degassing BC limitations, simulations considering the headspace were completed (MFR and SM approach) and showed significant improvement for the high-power (> 6,000 W m<sup>-3</sup>) Rushton simulations. All three-phase simulations utilize the headspace to ensure holdup accuracy and a realistic boundary condition for the oil. The degassing BC should not be applied for liquid dispersed phases as they only see the BC as an exit, which can create an erroneous mass imbalance. To further reduce the CFD and experimental gap, a 0.5 TSF was utilized for the high-power headspace simulations and showed slightly overestimated but reasonable gas holdup values compared to the experimental results (similar to the three-phase results in Figure 4C). Looking at relevant power inputs required for adequate hydrophobic feedstock fermentation mixing (> 6000 W  $m^{3-1}$ ), the 1200-RPM PBI setup showed increased gas holdup around the shaft and top two impellers with flow recirculation loops down the sides. This upper impeller's superior gas utilization efficiency is due to small air bubbles recirculating without moving through the impeller disc region.<sup>[43]</sup> This advantageous air recirculation flow pattern should help improve three-phase oil mixing and provide energy efficiency by producing similar gas holdup values with less power input (P/V), an essential scale-up factor.

## 3.3 | Three-phase (air, oil and water) volumetric mass transfer

The simplified two-phase simulations, either air-water or oil-water, point to liquid-liquid mixing advantages and more efficient gas distribution with the PBI setup. Three-phase (air, oil, and water) simulations with the headspace were then completed with a reduced 0.5 TSF for the higher 1000- and 1200-RPM stirring speeds for both the MFR and SM approaches. The SM approach showed poor CFD-experimental

# Biotechnology

18607314, 2024, 2, Downloaded from https://analyticalscien

onlinelibrary.wiley

.com/doi/10.1002/biot.202300384 by Qichao Yao - Johns Hopkins University, Wiley Online Library on [25/06/2025]. See the Terms and Condition

(https

. wiley

Wiley Online Library for

rules of use; OA articles are governed by the applicable Creative Commons

License

correlation, indicating future work should tune parameters impacting air coalescence in these high-power input fermenter conditions. The MFR method showed better experimental correlation and will be the only method discussed further. With three phases, the PBI setup showed improved mixing with 10% higher oil uniformity, as seen visually in Figure 5D. The added sparging-induced turbulence helps further distribute the oil away from the shaft and impellers. Liquidliquid mixing remains challenging to quantify, but the PBI setup at a 1200-RPM stirring speed pulls ~84% of the oil down into the original liquid domain (compared to only 63% for the 3R impeller setup at 1200-RPM stirring speeds), which we interpret as a general indicator of good fermentation mixing. The 3R impeller setup has large amounts of oil driven to the bottom and into the headspace region, which may reduce the accessibility of the yeast cells to the oil substrate and lead to decreased productivity in the oil fermentation.<sup>[5]</sup> Three-phase simulation gas holdup follows the same flow patterns as the two-phase simulations, as air and water are the dominant physics mechanisms. Including the headspace generally increased the holdup values and showed improved experimental correlation for the 1000 and 1200 RPM simulations (Figure 4C). Although the Rushton simulations feature slightly higher holdup values, the air is primarily grouped around the shaft and under the impellers, which limits access. The PBI setup generally shows more air circulating outside this impeller diameter area, which we believe benefits the fermentations (Figure 5H).

Figure 4D displays the CFD estimated  $k_l a$  values for the MFR three-phase simulations, which show good agreement with experimental results. 500–750 h<sup>-1</sup>  $k_l a$  values are well established for this type of bench-scale process at 1200-RPM stirring speeds similar to industry-scale mass transfer characteristics, typically corresponding to 100-150 mmol L<sup>-1</sup> h<sup>-1</sup> oxygen transfer rates. In addition, the 500-RPM stirring speed simulations show a reasonable  $k_l a$  range (117–143 h<sup>-1</sup>) for this type of process.<sup>[44-46]</sup> Slight increases compared to previous studies were expected due to the use of 2L-sized impellers in the 1L reactor, which leads to higher P/V values. In addition, PBI setups have previously demonstrated higher  $k_L a$  values with reduced power<sup>[47]</sup> with good mass transfer and flow field scalability from the laboratory to pilot-scale when compared to 3R impeller setups.<sup>[48]</sup> These trends correlate with improved Y. lipolytica fermentation with oil substrate under these conditions.<sup>[5]</sup> Overestimation of  $k_L a$  by the single bubble size was demonstrated previously<sup>[26]</sup> and is accounted for with  $C_k$ . The single bubble size used for the air phase does not account for bubble-bubble interactions and bubble turbulence as captured by the population balance equations, impacting multiple parameters. Viscosity also affects mass transfer and may need to be considered separately<sup>[21]</sup> as it impacts droplet/bubble size in population balances,<sup>[49]</sup> which would impact mass transfer.

Having demonstrated the PBI setup's mixing and energy efficiency benefits, it was desired to understand how lower (0.5 and 0.75 vvm) and higher (1.5 and 2.0 vvm) aeration rates impacted the 1200 RPM PBI setup CFD model. The lower aeration rates reduced holdup by up to 37% (with a similar  $k_L a$  reduction) while maintaining similar mixing characteristics and oil distribution. The high vvm simulations produced up to a 25% increase in holdup (with a similar  $k_L a$ 



**FIGURE 5** Oil volume fraction contours for the 3R (A and B) and PBI (C & D) impeller setups for the three-phase (air, oil, and water), 1000- and 1200-RPM simulations (most relevant for high-cell density fermentation with Yarrowia lipolytica). Air volume fraction contours for the 3R (E &F) and (G&H). Note: headspaces are not included for easier comparison to two-phase simulations. The liquid volume was set at 1 L for all simulations.

increase), which seemed promising initially, but a more detailed analysis revealed that ~90% of the additional air was bunched around the shaft and impellers. As previously discussed with the 3R setup, this bunching negatively impacts the oil mixing, seen in an oil uniformity reduction (~10%). Given the lack of additional, accessible air and reduced oil mixing, these aeration modifications do not look to provide measurable production benefits but need to be experimentally verified.

### 4 CONCLUDING REMARKS

A three-phase CFD model has been used to study a highly agitated fermenter's mass transfer and mixing characteristics utilizing hydrophobic feedstock such as plant oils. The three-phase (air, oil, and water) model was able to use both air bubble size and solution data to improve convergence and provide highly agreeable  $k_La$  values with experimental and literature data. The pitched-blade impeller setup at higher stirring speeds improved mixing characteristics with similar mass transfer while using reduced power compared to Rushton impeller setups. This reduced-power setup could have significant monetary implications at the industrial scale and certainly deserves further investigation.

### NOMENCLATURE

### **Greek Symbols**

- $\alpha$  Volume fraction
- **Γ** Torque
- $\xi$  Dimensionless size of eddies in the intertial subrange of turbulence
- ε Turbulence dissipation rate
- φ Uniformity index
- $\rho$  Density of phase
- μ Viscosity
- $\mu_{eff}$  Effective viscosity accounting for turbulence
- σ Surface tension coefficient (Air/Water = .072 and Oil/Water = .050  $\frac{N}{-}$ )
- $\sigma_k$  Turbulent Prandtl Number-kinetic energy (1.00)
- $\sigma_{\varepsilon}$  Turbulent Prandtl number -kinetic energy dissipation (1.30)
- u<sub>i</sub> Mean velocity
- $\overline{\tau_i}$  Reynolds stress tensor
- $\tau_{ii}$  Actual collision contact time
- $\omega$  Rotational speed (rev/s)
- ag Interfacial surface area from the predicted bubble size
- A<sub>i</sub> Area of the surface
- B<sub>K</sub> Break-up kernel

- BC Boundary condition
- C Constants
- C<sub>D</sub> Drag coefficient
- C<sub>k</sub> Volumetric mass transfer coefficient constant
- $C_{\varepsilon 1}$  Reynolds stress model constant (1.45)
- $C_{\varepsilon 2}$  Reynolds stress model constant (1.9)
- C<sub>o</sub> Actual dissolved oxygen concentration
- C<sub>o</sub><sup>\*</sup> Saturated dissolved oxygen concentration
- $c_{\mu}$  k-  $\varepsilon$  turbulence model constant (.09)
- D Diffusion coefficient of oxygen in water (1.98e-9  $\frac{m^2}{2}$ )
- $D_B$  Death rate of a dispersed phase due to break-up
- D<sub>C</sub> Death rate of a dispersed phase to coalescence
- D<sub>I</sub> Impeller diameter (m)
- D<sub>T</sub> Tank diameter
- d Bubble diameter
- d<sub>bg</sub> Local Sauter mean bubble diameter
- $d_i$  Diameter of bubble/droplet in bin i (or j)
- E<sub>o</sub> Eötvös number
- $f_i$  Size group fraction of the *i*th bubble group
- $f_{BV}$  Break-up fraction (dimensionless)
- F Calibration coefficient
- $\vec{F_i}$  Coriolis and centrifugal forces
- $\vec{g}$  Gravitational acceleration constant
- *h*<sub>f</sub> Critical rupture thickness
- h<sub>0</sub> Initial film thickness
- H Empirical M and  $E_o$  functions
- J Empirical M and E<sub>o</sub> functions
- k Turbulence kinetic energy
- k<sub>L</sub> Liquid mass transfer resistance
- k<sub>L</sub>a Volumetric mass transfer coefficient
- *K<sub>i</sub>* Exchange coefficient of liquid and polydispersed phases (oil and water)
- I Liquid phase (water)
- m Mass
- *m* Mass flowrate
- M Morton number
- $M_i$  Interphase momentum exchange term
- MUSIG
  - N<sub>P</sub> Power number
  - o Oil phase (corn oil)

Multiple size group

- $n_i$  Density of particles of mass m at time t
- $(\Delta_P)_i$  Pressure difference around the impeller at surface i
  - p Pressure
  - P Power
- $P_B$  Production rate of a dispersed phase due to break-up
- $P_C$  Production rate of a dispersed phase to coalescence
- $P_l$  Turbulent kinetic energy due to shear
- PBM Population balance modeling
- OUR Oxygen uptake rate
- OTR Oxygen transfer rate
  - $r_i$  Radial distance from the axis of the mounted impeller shaft
  - r<sub>ij</sub> Equivalent radius

- Re Reynolds number
- $\vec{R}_i$  Interfacial momentum
- RMS Root mean square
- S<sub>i</sub> Source terms
- SM Sliding mesh
- t Time
- $t_{ij}$  Time required for coalescence between particle i and j
- TSF Time scale factor
- u Velocity
- U<sub>T</sub> Terminal bubble velocity
- v Mass volume fraction of size group i
- V Reactor volume
- vvm Gas sparging rate (volume/min) per unit volume of unaerated liquid
- W<sub>1</sub> Impeller width (vertical distance)
- W<sub>B</sub> Width of the baffle (From OD to ID)
- c Continuous phase
- dp Dispersed phase (oil or gas)
- e Experimentally estimated
- i Mother particle to be broken into smaller (daughter particles)
- j Daughter particle originating from break-up of larger (mother) particle
- I Liquid phase (continuous phase, water)
- g Gas phase (dispersed phase, air bubbles)
- o Oil phase (polydispersed phase, oil droplets)
- PT Higbie's penetration theory
- P Power
- r Location vector
- R Lubrication
- t Turbulent

### Superscripts

- drag Drag
- B Buoyancy
- P Dispersion forces

### AUTHOR CONTRIBUTIONS

Richard Marx: Data curation (lead); formal analysis (lead); investigation (lead); methodology (lead); project administration (lead); resources (lead); software (lead); validation (lead); visualization (lead); writing original draft (lead); writing—review and editing (lead). Huolong Liu: Data curation (supporting); formal analysis (supporting); investigation (supporting); methodology (supporting); writing—original draft (supporting); writing—review and editing (supporting). Seongkyu Yoon: Conceptualization (supporting); funding acquisition (supporting); project administration (supporting); resources (supporting); writing original draft (supporting); writing—review and editing (supporting). Dongming Xie: Conceptualization (lead); data curation (supporting); formal analysis (supporting); funding acquisition (lead); investigation (supporting); project administration (lead); supervision (lead); writing—original draft (supporting); writing—review and editing (lead). 12 of 14

### Biotechnology

Journal

### ACKNOWLEDGMENTS

This project was supported by the National Science Foundation (#1911480). The authors thank Dr. Iman Mirzaee for providing CFD simulation feedback and suggestions. In addition, we would like to thank Mr. Jiansong Qin and Ms. Elif Kurt for helping run the 1-L bioreactor experiments.

### CONFLICT OF INTEREST STATEMENT

The authors declare no conflicts of interest.

### DATA AVAILABILITY STATEMENT

The data that support the findings of this study are available from the corresponding author upon reasonable request.

### ORCID

### REFERENCES

- 1. Zieniuk, B., & Fabiszewska, A. (2018). Yarrowia lipolytica: A beneficious yeast in biotechnology as a rare opportunistic fungal pathogen: a minireview. World Journal of Microbiology and Biotechnology, 35, 10.
- Xie, D., Jackson, E., & Zhu, Q. (2015). Sustainable source of omega-3 eicosapentaenoic acid from metabolically engineered *Yarrowia lipolytica* : From fundamental research to commercial production. *Applied Microbiology & Biotechnology*, *99*, 1599–1610.
- Liu, N., Liu, B., Wang, G., Soong, Y. H. V., Yong, T., Weifeng, L., & Dongming, X. (2020). Lycopene production from glucose, fatty acid and waste cooking oil by metabolically engineered *Escherichia coli*. *Biochemical Engineering Journal*, 155, 107488.
- Soong, Y. H. V., Zhao, L., Liu, N., Yu, P., Lopez, C., Olson, A., Wong, H. W., Shao, Z., & Xie, D. (2021). Microbial synthesis of wax esters. *Metabolic Engineering*, 67, 428.
- Liu, N., Soong, Y. H. V., Mirzaee, I., Olsen, A., Yu, P., Wong, H. W., & Xie, D. (2021). Biomanufacturing of value-added products from oils or fats: A case study on cellular and fermentation engineering of *Yarrowia lipolytica*. *Biotechnology and Bioengineering*, 118, 1677.
- Xie, D. (2017). Integrating cellular and bioprocess engineering in the non-conventional yeast Yarrowia lipolytica for biodiesel production: A review. Frontiers in Bioengineering and Biotechnology, 5, 65.
- Kamla, Y., Bouzit, M., Hadjeb, A., Arab, I. M., & Beloudane, M. (2016). CFD study of the effect of baffles on the energy consumption and the flow structure in a vessel stirred by a Rushton turbine. *Mechanika*, 22, 190.
- Matran, R. M., Galaction, A.-I., Blaga, A. C., Turnea, M., & Caşcaval, D. (2016). Distribution of mixing efficiency in a split-cylinder gas-lift bioreactor with immobilized *Yarrowia lipolytica* cells used for olive oil mill wastewater treatment. *Chemical Engineering Communications*, 203, 666–675.
- Puthli, M. S., Rathod, V. K., & Pandit, A. B. (2005). Gas-liquid mass transfer studies with triple impeller system on a laboratory scale bioreactor. *Biochemical Engineering Journal*, 23, 25–30.
- Gogate, P. R., Beenackers, A. A. C. M., & Pandit, A. B. (2000). Multipleimpeller systems with a special emphasis on bioreactors: A critical review. *Biochemical Engineering Journal*, 6, 109–144.
- Luo, H., & Svendsen, H. F. (1996). Theoretical model for drop and bubble breakup in turbulent dispersions. AIChE Journal, 42, 1225–1233.
- Naeeni, S. K., & Pakzad, L. (2019). Droplet size distribution and mixing hydrodynamics in a liquid–liquid stirred tank by CFD modeling. *International Journal of Multiphase Flow*, 120, 103100.

- Laakkonen, M., Alopaeus, V., & Aittamaa, J. (2006). Validation of bubble breakage, coalescence and mass transfer models for gas-liquid dispersion in agitated vessel. *Chemical Engineering Communications*, 61, 218–228.
- Gelves, R., Dietrich, A., & Takors, R. (2014). Modeling of gas-liquid mass transfer in a stirred tank bioreactor agitated by a Rushton turbine or a new pitched blade impeller. *Bioprocess and Biosystems Engineering*, 37, 365.
- Wang, H., Jia, X., Wang, X., Zhou, Z., Wen, J., & Zhang, J. (2014). CFD modeling of hydrodynamic characteristics of a gas-liquid two-phase stirred tank. *Applied Mathematical Modelling*, 38, 63–92.
- Xie, R., Li, J., Jin, Y., Zou, D., & Chen, M. (2019). Simulation of drop breakage in liquid-liquid system by coupling of CFD and PBM: Comparison of breakage kernels and effects of agitator configurations. *Chinese Journal of Chemical Engineering*, 27, 1001–1014.
- Sarhan, A. R., Naser, J., & Brooks, G. (2017). CFD modeling of threephase flotation column incorporating a population balance model. *Procedia Engineering*, 184, 313–317.
- McClure, D. D., Kavanagh, J. M., Fletcher, D. F., & Barton, G. W. (2015). Oxygen transfer in bubble columns at industrially relevant superficial velocities: Experimental work and CFD modelling. *Chemical Engineering Journal*, 280, 138–146.
- Laakkonen, M., Moilanen, P., Alopaeus, V., & Aittamaa, J. (2007). Modelling local bubble size distributions in agitated vessels. *Chemical Engineering Science*, 62, 721–740.
- Bach, C., Yang, J., Larsson, H., Stocks, S. M., Gernaey, K. V., Albaek, M. O., & Krühne, U. (2017). Evaluation of mixing and mass transfer in a stirred pilot scale bioreactor utilizing CFD. *Chemical Engineering Science*, 171, 19–26.
- Brucato, A., Ciofalo, M., Grisafi, F., & Micale, G. (1998). Numerical prediction of flow fields in baffled stirred vessels: A comparison of alternative modelling approaches. *Chemical Engineering Science*, 53, 3653–3684.
- Murthy, B. N., Ghadge, R. S., & Joshi, J. B. (2007). CFD simulations of gas-liquid-solid stirred reactor: Prediction of critical impeller speed for solid suspension. *Chemical Engineering Science*, 62, 7184–7195.
- Ahmed, S. U., Ranganathan, P., Pandey, A., & Sivaraman, S. (2010). Computational fluid dynamics modeling of gas dispersion in multi impeller bioreactor. *Journal of Bioscience, & Bioengineering*, 109, 588–597.
- Gentric, C., Mignon, D., Bousquet, J., & Tanguy, P. A. (2005). Comparison of mixing in two industrial gas-liquid reactors using CFD simulations. *Chemical Engineering Science*, 60, 2253–2272.
- Jian, M., Yuyun, B., Lei, C., John, M. S., & Zhengming, G. (2008). Numerical simulation of gas dispersion in an aerated stirred reactor with multiple impellers. *Industrial & Engineering Chemistry Research*, 47, 7112–7117.
- Kerdouss, F., Bannari, A., Proulx, P., Bannari, R., Skrga, M., & Labrecque, Y. (2008). Two-phase mass transfer coefficient prediction in stirred vessel with a CFD model. *Computers and Chemical Engineering*, 32, 1943–1955.
- Sarkar, J., Kanwar, L., Loomba, V., & Rathore, A. (2016). CFD of mixing of multi-phase flow in a bioreactor using population balance model. *Biotechnology Progress*, 32, n/a-n/a.
- Hong, H. S. (2014). Simulation of gas-inducing reactor couples gasliquid mass transfer and biochemical reaction. *Biochemical Engineering Journal*, 91, 1-9-2014 v.2091.
- Hsu, R. C., Chiu, C. K., & Lin, S. C. (2018). A CFD study of the drawdown speed of floating solids in a stirred vessel. *Journal of the Taiwan Institute* of Chemical Engineers, 90, 33–43.
- González-Sáiz, J.-M., Garrido-Vidal, D., & Pizarro, C. (2009). Scale up and design of processes in aerated-stirred fermenters for the industrial production of vinegar. *Journal of Food Engineering*, 93, 89–100.
- Bao, Y., Huang, X., Shi, L., & Wang, Y. (2002). Mechanism of off-bottom suspension of solid particles in a mechanical stirred tank. *Chinese Journal of Chemical Engineering*, 10, 476–479.

- Bao, Y., Hao, Z., Gao, Z., Shi, L., & Smith, J. M. (2005). Suspension of buoyant particles in a three phase stirred tank. *Chemical Engineering Science*, 60, 2283.
- Julian Becker, P., Puel, F., Jakobsen, H. A., & Sheibat-Othman, N. (2014). Development of an improved breakage kernel for high dispersed viscosity phase emulsification. *Chemical Engineering Science*, 109, 326–338.
- Cheung, S. C. P., Yeoh, G. H., & Tu, J. Y. (2007). On the modelling of population balance in isothermal vertical bubbly flows--Average bubble number density approach. *Chemical Engineering*, & *Processing: Process Intensification*, 46, 742–756.
- 35. Montoya, G., Sanyal, J., Braun, M., & Azhar, M. (2019). On the assessment, implementation, validation, and verification of drag and lift forces in gas-liquid applications for the CFD codes FLUENT and CFX. Experimental and Computational Multiphase Flow, 1, 255–270.
- Frank, T., Shi, J., & Burns, A. D. (2004). Proceeding of the third international symposium on two-phase modelling and experimentation, Citeseer, 22–25.
- Xie, M., Xia, J., Zhou, Z., Chu, J., Zhuang, Y., & Zhang, S. (2014). Flow pattern, mixing, gas holdup and mass transfer coefficient of triple-impeller configurations in stirred tank bioreactors. *Industrial & Engineering Chemistry Research*, 53, 5941–5953.
- Jamshidian, R., Scully, J., & Van den Akker, H. E. A. (2023). Two-fluid simulations of an aerated lab-scale bioreactor. *Chemical Engineering Research & Design: Transactions of the Institution of Chemical Engineers Part A*, 196, 254–275.
- Lane, G. L., Schwarz, M. P., & Evans, G. M. (2005). Numerical modelling of gas–liquid flow in stirred tanks. *Chemical Engineering Science*, 60, 2203–2214.
- 40. Wang, H., Gao, Z., Wang, B., Bao, Y., & Cai, Z. (2020). Gas dispersion and solid suspension in a three-phase stirred tank with triple impellers. *Chinese Journal of Chemical Engineering*, 28, 1195–1202.
- 41. Fujasová, M., Linek, V., & Moucha, T. (2007). Mass transfer correlations for multiple-impeller gas-liquid contactors. Analysis of the effect of axial dispersion in gas and liquid phases on "local" kLa values measured by the dynamic pressure method in individual stages of the vessel. *Chemical Engineering Science*, 62, 1650–1669.
- Arjunwadkar, S. J., Saravanan, K., Pandit, A. B., & Kulkarni, P. R. (1998). Optimizing the impeller combination for maximum hold-up with minimum power consumption. *Biochemical Engineering Journal*, 1, 25–30.
- Bernauer, S., Schöpf, M., Eibl, P., Witz, C., Khinast, J., & Hardiman, T. (2021). Characterization of the gas dispersion behavior of multiple impeller stages by flow regime analysis and CFD simulations. *Biotechnology and Bioengineering*, 118, 3058.
- 44. Ferreira, P., Lopes, M., Mota, M., & Belo, I. (2016). Oxygen mass transfer impact on citric acid production by *Yarrowia lipolytica* from crude glycerol. *Biochemical Engineering Journal*, 110, 35–42.
- Palmerín-Carreño, D. M., Castillo-Araiza, C. O., Rutiaga-Quiñones, O. M., Verde-Calvo, J. R., & Huerta-Ochoa, S. (2016). Kinetic, oxygen mass transfer and hydrodynamic studies in a three-phase stirred tank bioreactor for the bioconversion of (+)-valencene on *Yarrowia lipolytica* 2.2ab. *Biochemical Engineering Journal*, 113, 37–46.
- Santos, A. G., Ribeiro, B. D., do Nascimento, F. V., & Coelho, M. A. Z. (2019). Culture miniaturization of lipase production by *Yarrowia lipolytica*. *Current Biochemical Engineering*, 5, 12–20.
- Lone, S. R., Kumar, V., Seay, J. R., Englert, D. L., & Hwang, H. T. (2020). Mass transfer and rheological characteristics in a stirred tank bioreactor for cultivation of *Escherichia coli* BL21. *Biotechnology and Bioprocess Engineering*, 25, 766.
- Guo, D. S., Ji, X. J., Ren, L. J., Li, G. L., Sun, X. M., Chen, K. Q., Gao, S., & Huang, H. (2018). Development of a scale-up strategy for fermentative production of docosahexaenoic acid by Schizochytrium sp. *Chemical Engineering Communications*, 176, 600–608.

- Abujelala, M. T., & Lilley, D. G. (1984). Limitations and empirical extensions of the k-ε model as applied to turbulent confined swirling flows. Chemical Engineering Communications, 31, 223–236.
- Olmos, E., Gentric, C., Vial, C., Wild, G., & Midoux, N. (2001). Numerical simulation of multiphase flow in bubble column reactors. Influence of bubble coalescence and break-up. *Chemical Engineering Science*, 56, 6359–6365.
- Khopkar, A. R., Kasat, G. R., Pandit, A. B., & Ranade, V. V. (2006). CFD simulation of mixing in tall gas–liquid stirred vessel: Role of local flow patterns. *Chemical Engineering Science*, 61, 2921–2929.
- McClure, D. D., Kavanagh, J. M., Fletcher, D. F., & Barton, G. W. (2014). Development of a CFD model of bubble column bioreactors: Part two

   comparison of experimental data and CFD predictions. *Chemical Engineering & Technology*, *37*, 131–140.
- Simonnet, M., Gentric, C., Olmos, E., & Midoux, N. (2008). CFD simulation of the flow field in a bubble column reactor: Importance of the drag force formulation to describe regime transitions. *Chemical Engineering & Processing*, 47, 1726–1737.
- Liao, Y. (2020). Update to the MUSIG model in ANSYS CFX for reliable modelling of bubble coalescence and breakup. *Applied Mathematical Modelling*, 81, 506–521.
- Vakili, M. H., & Esfahany, M. N. (2009). CFD analysis of turbulence in a baffled stirred tank, a three-compartment model. *Chemical Engineering Science*, 64, 351–362.
- Andersson, R., & Andersson, B. (2006). On the breakup of fluid particles in turbulent flows. *AIChE Journal*, 52, 2020–2030.
- Liu, H., & Li, M. (2014). Two-compartmental population balance modeling of a pulsed spray fluidized bed granulation based on computational fluid dynamics (CFD) analysis. *International Journal of Pharmaceutics*, 475, 256–269.
- Liu, H., Yoon, S., & Li, M. (2017). Three-dimensional computational fluid dynamics (CFD) study of the gas-particle circulation pattern within a fluidized bed granulator: By full factorial design of fluidization velocity and particle size. *Drying Technology*, 35, 1043–1058.
- Khalil, A., Rosso, D., & DeGroot, C. T. (2021). Effects of flow velocity and bubble size distribution on oxygen mass transfer in bubble column reactors--A critical evaluation of the computational fluid dynamicspopulation balance model. *Water Environment Research*, 10614303, 1.
- Bhole, M. R., Joshi, J. B., & Ramkrishna, D. (2008). CFD simulation of bubble columns incorporating population balance modeling. *Chemical Engineering Science*, 63, 2267–2282.
- Zadghaffari, R., Moghaddas, J. S., & Revstedt, J. (2010). Large-eddy simulation of turbulent flow in a stirred tank driven by a Rushton turbine. Computers and Fluids, 39, 1183–1190.
- Sun, K., Zhao, H., Zhao, K., Li, D., & Bai, S. (2020). Optimization of SCR inflow uniformity based on CFD simulation. *Open Physics*, 18, 1168– 1177.

### SUPPORTING INFORMATION

Additional supporting information can be found online in the Supporting Information section at the end of this article.

How to cite this article: Marx, R., Liu, H., Yoon, S., & Xie, D. (2024). CFD evaluation of hydrophobic feedstock bench-scale fermenters for efficient high agitation volumetric mass transfer. *Biotechnology Journal*, *19*, e2300384. https://doi.org/10.1002/biot.202300384